

Vitamin B₁₂ and protein biosynthesis. V. The site of action of vitamin B₁₂ and its inhibition by a B₁₂ antagonist*,**

It has been reported from this laboratory that vitamin B₁₂ functions *in vivo* and *in vitro* in the incorporation of amino acids into protein^{1,2,3,4}, when the microsomal system of KELLER AND ZAMEČNIK⁵ is used. It was found³ that much of the vitamin B₁₂ is present in the pH 5 enzymes. These enzymes have now been fractionated for the separation of the vitamin B₁₂-containing enzyme(s). For this purpose 6.0 μ C of ⁶⁰Co-labeled vitamin B₁₂ were injected into a rat, and 8 h later the rat was sacrificed and pH 5 enzymes prepared from the liver by the method of KELLER AND ZAMEČNIK⁵. The nucleic acids present in these enzymes were removed by protamine sulfate precipitation. Ammonium sulfate fractionation of the resultant solution showed that essentially all of the radioactivity (*i.e.*, the vitamin B₁₂) is present in the fraction precipitating between 40 and 60% saturation (Table I).

TABLE I
DISTRIBUTION OF RADIOACTIVE B₁₂ IN AMMONIUM SULFATE FRACTIONS

	Counts/min	Counts/min/mg
Total radioactivity in 105,000 \times g supernatant	88600	
Total radioactivity in pH 5.0 enzyme	67400	
Total radioactivity after nucleic acid separation	62000	
Total radioactivity in 0-20% ammonium sulfate fraction	Nil	
Total radioactivity in 20-40% ammonium sulfate fraction	6800	2130
Total radioactivity in 40-60% ammonium sulfate fraction	51800	17250
Total radioactivity in 60-80% ammonium sulfate fraction	Nil	
Total radioactivity in 80-100% ammonium sulfate fraction	Nil	

TABLE II
INHIBITION AND INCORPORATION OF 2-¹⁴C-ALANINE
INTO PROTEINS BY MICROSOME PREPARATIONS SUPPLEMENTED WITH pH 5 ENZYMES,
AMMONIUM SULFATE FRACTIONS OF pH 5 ENZYMES AND ANTI B₁₂

Complete system contained 0.3 ml microsomal suspension, 0.2 ml pH 5 enzymes solution; 0.25 μ mole 2-¹⁴C-alanine; 10 μ moles phospho(enol)pyruvate; 0.02 mg pyruvate kinase; 0.5 μ mole ATP and 0.25 μ mole GTP. The incubation mixtures 5-12 contained same amount of 2-¹⁴C-alanine, phospho(enol)pyruvate, pyruvate kinase, ATP and GTP. The final volume of the incubation mixture was 1.0 ml. MS = microsomes; NA = supernatant nucleic acid; ASF = ammonium sulfate fraction.

System used	Counts min. mg protein
1. Complete system	93
2. Complete system minus microsomes	32
3. Complete system minus pH 5 enzymes	21
4. Complete system + anti B ₁₂ (250 μ g)	52
5. 0.3 ml MS + 0.5 mg NA + 2 mg ASF (0-20%)	24
6. 0.3 ml MS + 0.5 mg NA + 2 mg ASF (20-40%)	67
7. 0.3 ml MS + 0.5 mg NA + 2 mg ASF (40-60%)	154
8. 0.3 ml MS + 0.5 mg NA + 2 mg ASF (60-80%)	28
9. 0.3 ml MS + 0.5 mg NA + 2 mg ASF (80-100%)	20
10. 0.3 ml MS + 0.5 mg NA + 1 mg ASF (20-40%) + 1 mg ASF (40-60%)	216
11. 0.3 ml MS + 0.5 mg NA + 1 mg ASF (20-40%) + 1 mg ASF (40-60%) + Anti B ₁₂ (250 μ g)	122
12. 0.3 ml MS + 0.5 mg NA + 2 mg ASF (20-40%) + 2 mg ASF (40-60%) + Anti B ₁₂ (250 μ g)	193

* Supported in part by grants-in-aid from the National Vitamin Foundation and the U.S. Atomic Energy Commission (AT(11-1)-67).

** The following abbreviations are used: ATP, adenosine triphosphate; AA, amino acid; AMP, adenosine monophosphate; PP, inorganic pyrophosphate; GTP, guanosine triphosphate.

The activity of the various ammonium sulfate fractions was studied with regard to their requirement (in place of the pH 5 enzymes) for the incorporation of amino acids into protein. Table II shows that the 40–60% fraction as well as other fractions (20–40% fraction) and nucleic acids are required.

In a search for the function of this vitamin B₁₂ enzyme (40–60% fraction) experiments were carried out on the initial step of the amino acid activation. In these experiments, the 40–60% fraction was incubated with ATP, amino acid and ³²P-labeled pyrophosphate. A large amount of radioactivity was found to be incorporated into ATP (total counts incorporated, 680/min). This was not the case when other ammonium sulfate fractions of the pH 5 enzymes were used in place of the 40–60% fraction. These results indicate that vitamin B₁₂ functions in the activation reaction, $AA + ATP \xrightarrow{\text{B}_{12} \text{ enzyme}} \sim AA + PP$. An antivitamin B₁₂ compound, the anilide of the monocarboxylic acid of vitamin B₁₂⁶, was used to test further the specificity of vitamin B₁₂ for this reaction. The inclusion of this antagonist in the incubation mixture, in an amount equal to 1000 times that of the vitamin, greatly decreased (55%) the incorporation of pyrophosphate into ATP (total counts incorporated, 310/min). This was partially reversed by doubling the vitamin B₁₂ enzyme (40–60% fraction).

This same antagonist was then used in the usual microsomal system for amino acid incorporation into protein. Table II shows that about 44% reduction in incorporation was obtained, and this reduction was reversed by the addition of a large amount of vitamin B₁₂ enzyme.

These results indicate that vitamin B₁₂ is bound to an enzyme (or enzymes) which is involved in the activation of amino acid for incorporation into protein.

We are grateful to Dr. C. ROSENBLUM of Merck & Co. Inc. for the generous gift of ⁶⁰Co-labeled vitamin B₁₂ and to Dr. E. LESTER SMITH of Glaxo Laboratories (England) for antivitamin B₁₂ compounds.

Department of Animal Science, Division of Animal Nutrition
University of Illinois, Urbana, Ill. (U.S.A.)

S. R. WAGLE
RANJAN MEHTA
B. CONNOR JOHNSON

¹ S. R. WAGLE AND B. CONNOR JOHNSON, *Arch. Biochem. Biophys.*, 70 (1957) 619.

² S. R. WAGLE, RANJAN MEHTA AND B. CONNOR JOHNSON, *J. Am. Chem. Soc.*, 79 (1957) 4249.

³ S. R. WAGLE, RANJAN MEHTA AND B. CONNOR JOHNSON, *Arch. Biochem. Biophys.*, 72 (1957) 241.

⁴ S. R. WAGLE, RANJAN MEHTA AND B. CONNOR JOHNSON, *J. Biol. Chem.*, 230 (1958) 137.

⁵ E. B. KELLER AND P. C. ZAMECNIK, *J. Biol. Chem.*, 221 (1956) 45.

⁶ E. LESTER SMITH, *Vitamin B₁₂ and Intrinsic Factor. I. Europäisches Symposium, Hamburg, 1956*, Ferdinand Enke Verlag, Stuttgart, 1957, pp. 1–9.

Received January 4th, 1958

Biosynthesis of ribose and desoxyribose in *Escherichia coli**

Tracer studies^{1,2,3} suggest that ribose arises largely from glucose, while the origin of desoxyribose is less certain. In the present work RNA, DNA and polyglucosan⁴ were isolated from *E. coli* R-2 adapted to grow on acetate as sole carbon source. RNA was converted to mononucleotides by KOH digestion, DNA was recovered by acid precipitation and degraded to mononucleotides⁵. Nucleotides were purified by ion-exchange chromatography^{6,7}.

Ribose. Purine ribotides were hydrolyzed with phosphatase⁸ and 1 N H₂SO₄. Cytidylic acid was deaminated and added to the original uridylic acid, which was then hydrolyzed with phosphatase and uridine nucleosidase. The ribose obtained was purified on cellulose columns and then degraded⁹.

Desoxyribose. The purine nucleotides were combined and hydrolyzed to desoxyribose-5-phosphate, which was then converted to acetaldehyde and lactic acid by incubation with extracts of *E. coli*¹⁰ and rabbit muscle. Acetaldehyde was oxidized to acetic acid and the acids were degraded⁹.

Glucose. The polyglucosan was isolated from a portion of the culture by the procedure used for mammalian glycogen, and hydrolyzed with 1 N H₂SO₄. Glucose was degraded by the method of BERNSTEIN *et al.*¹¹.

All samples were converted to BaCO₃ and radioactivity measurements carried out in a windowless counter, and corrected to infinite thinness. The results are shown in Table I. In experiments with 22-h cultures, the isotope distribution in ribose and desoxyribose was almost identical with that shown in the Table, and the purine- and pyrimidine-bound ribose had the same distribution.

* Supported in part by the U.S. Atomic Energy Commission, Contract AT-(30-1)-1320.